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$[\text{Co}(\text{H}_2\text{O})_6]^{2+}(\text{aq}) + 4\text{Cl}^{-}(\text{aq}) \rightleftharpoons [\text{CoCl}_4]^{2-}(\text{aq}) + 6\text{H}_2\text{O}(\text{l})$. In many given examples of this system, the addition of $\text{H}_2\text{O}(\text{l})$ is shown to create a pink color (as indeed it does in Expt. 17 of the first edition of the Vonderbrink lab manual) and then is accompanied by an explanation of the addition of a product and in terms of Le Châtelier's principle. One might assume that since ...

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