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Corrosion, wearing away due to
chemical reactions, mainly oxidation
(see oxidation-reduction, oxide). It
occurs whenever a gas or liquid
chemically attacks an exposed
surface, often a metal, and is

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accelerated by warm temperatures
and by acids and salts. Normally,
corrosion products (e.g., rust,

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Crevice corrosion -- corrosion that occurs in stagnant locations such as those found under gaskets. Filiform corrosion -- corrosion that occurs when water gets under a coating such as paint. Galvanic Corrosion: This can occur when two different metals are located together in a liquid

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What Is Metal Corrosion and Why
Does It Occur?

For most of us, corrosion is the
degradation of metals that is often
called rust and is a curse we have to

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endure helplessly. The mission of this popular Web site is to improve the general awareness of what metal corrosion is all about, from complicated processes that are at play even behind the simplest corrosion reactions to the impact corrosion may have on complex

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Corrosion science and engineering
information hub

Corrosion prevention can take a
number of forms depending on the
circumstances of the metal being

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corroded. Corrosion prevention techniques can be generally classified into 6 groups: Environmental Modification . Corrosion is caused by chemical interactions between metal and gasses in the surrounding environment. By removing the metal from, or ...

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Getting your medicine or a
prescription. You may be able to get
your medicine or a prescription in one
of the following ways: seeing a local

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GP and asking for a prescription. Staff at an NHS walk-in centre may be able to arrange for you to see a GP; asking a local pharmacist if they can provide an emergency supply of your medicine

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