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ISBN1-891389-31-9(alk.paper)

1.Chemistry,Physicalorganic.

I.Dougherty,DennisA.,1952- II.Title.

QD476.A57 2004 547 .13-dc22 2004049617

PrintedintheUnitedStatesofAmerica 09 08 07 06

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The reaction constants ρ_H and ρ_S derived from the dependences of the ρ_H and ρ_S activation parameters on substituent constants σ in S_NAr reactions of substituted benzenes with anionic and neutral nucleophiles in various solvents were found to be linearly related upon variation of substituents in the substrate and nucleophile.

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