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~~Endothermic? Hess's Law 15.1 Construct a Born-Haber cycle for group 1 and 2 oxides and chlorides [HL IB Chemistry] How to Calculate Enthalpy of Combustion - Mr Pauller Hess Law | Problem Solving | Heat of Formation | Introduction to Bond Energies (enthalpies) 5.3 Average Bond Enthalpy Calculations [SL IB Chemistry]~~

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Pogil Answer Key Chemistry Bond Energy

The energies you calculated in Questions 5c and 6a above are called bond energies. The bond energy for a particular type of bond can vary from one molecule to another because the atomic environment of a bond can influence the amount of energy needed to break the bond.

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